

LC-MS & UHPLC-MS solvents

The right purity for an excellent result







LC-MS and UHPLC-MS combine the advantages of chromatographic separation with those of Mass Spectrometry. UHPLC has the added advantage of being faster (fivefold or higher raw throughput) and having a lower solvent consumption (at least 90%) compared to conventional HPLC. When it comes to the detection, the structural information of the separated compounds given by Mass Spectrometry allows their identification and quantification, even if they are in complex matrixes. These features make both techniques widely used for pharmaceutical QC, synthesis of organic compounds and environmental analysis, as well as for genomics and proteomics.

The transition from liquid chromatography with UV detection to chromatography with MS detection is not trivial. It is necessary to adapt the methods to the new analytical conditions in order to obtain the maximum benefits associated to LC-MS/UHPLC-MS.

The efficiency of a LC-MS or an UHPLC-MS instrument is increased with the use of cleaner eluents. In this context, an HPLC grade solvent is not the most suitable option for its use in LC-MS/UHPLC- MS, as some impurities that are not detected by UV spectroscopy could interfere in the Mass Spectrometry. In addition, the use of LC-MS or UHPLC-MS quality solvents avoids equipment obstructions and reduces maintenance.

To reduce the amount of undesired impurities even more, all our LC-MS or UHPLC-MS bottles undergo a special treatment to avoid metal migration from the glass to the solvent.

Our solvents and mixtures for LC-MS and our solvents for UHPLC-MS respond to these requirements, ensuring:

- LOW CONTENT OF METALLIC IMPURITIES
- MICROFILTERED THROUGH 0,1 μm
- MINIMUM LEVEL OF ACIDITY AND ALKALINITY

 IMPURITIES CONTROLLED BY LC-MS OR UHPLC-MS (RESERPINE TEST DONE

LOW LEVEL OF NON-VOLATILE IMPURITIES

CERTIFICATE OF ACTUAL BATCH ANALYSIS

ON EACH BATCH)

LOW WATER CONTENT



What will you gain by using our LC-MS solvents and mixtures or our UHPLC-MS solvents?

- · Simpler and cleaner spectra
- · Prevent the formation of unwished adducts with metallic impurities
- · Longer column lifetimes
- · Avoid equipment obstructions
- · Savings in equipment maintenance



Metallic impurities at the ppm level that do not affect a conventional HPLC quantification, can distort its mass spectrum, modifying the abundance of the molecular ions of interest and complicating its interpretation. Below two spectra of the same peptide are shown, where the effect of using acetonitrile and water of LC-MS quality is clearly observed. The analysed peptide is human gastrin and the major ion, $[M+2H]^{2+}$, should appear at m/z = 1050.

In Fig. 1, the desired peak is masked by other peaks corresponding to adducts from the peptide with other alkaline metals ($[M+Na+H]^{2+}$, m/z = 1063 or $[M+K+H]^{2+}$, m/z = 1070), when using an HPLC grade solvent. As contrast, Fig. 2 shows a clear and neat peak corresponding to the desired $[M+2H]^{2+}$ ion, when a Scharlau's acetonitrile and water LC-MS solvent was employed.





Figure 1.

HG spectrum obtained with HPLC grade acetonitrile and water.

Figure 2.

HG spectrum obtained with Scharlau's LC-MS acetonitrile and water.

LC-MS analytical conditions

- \bullet Eluent: ACN/H $_{\rm 2}$ 0 mixture, 50/50, v/v, with 0,2% formic acid
- Flux: 250 µl/min. Split.
- Injection volume: 50 μl of a 10 $\mu g/m l$ human gastrin solution
- Detection: ESI + Frag. 3500 V Source T: 150 °C

Ordering information

UHPLC-MS solvents

| Description | Reference |
|------------------------|-----------|
| Acetonitrile, UHPLC-MS | AC0391 |
| Methanol, UHPLC-MS | ME0334 |
| Water, UHPLC-MS | AG0015 |

LC-MS solvents

| Description | Reference |
|----------------------|-----------|
| Acetonitrile, LC-MS | AC0371 |
| Ethyl acetate, LC-MS | AC0158 |
| Methanol, LC-MS | ME0326 |
| 2-Propanol, LC-MS | AL0326 |
| Water, LC-MS | AG0006 |

LC-MS blends

Scharlau offers ready-to-use solvent/modifier mixtures, greatly simplifying eluent preparation and assuring its suitability for LC-MS analysis.

| Description | Reference |
|--|-----------|
| Acetonitrile with 0,1% acetic acid, LC-MS | AC0374 |
| Acetonitrile with 0,1% formic acid, LC-MS | AC0373 |
| Acetonitrile with 0,1% trifluoroacetic acid, LC-MS | AC0372 |
| Methanol with 0,1% acetic acid, LC-MS | ME0329 |
| Methanol with 0,1% ammonium acetate, LC-MS | ME0330 |
| Methanol with 0,1% trifluoroacetic acid, LC-MS | ME0327 |
| Water with 0,1% acetic acid, LC-MS | AG0009 |
| Water with 0,1% ammonium acetate, LC-MS | AG0010 |
| Water with 0,1% trifluoroacetic acid, LC-MS | AG0007 |
| Water with 0,1% formic acid, LC-MS | AG0008 |

LC-MS additives

In the preparation of eluents for LC-MS it is common to add modifiers in order to promote the formation of molecular ions, thus improving spectral peak shapes.

| Description | Reference |
|--|-----------|
| Acetic acid glacial, eluent additive for LC-MS | AC0347 |
| Ammonia, solution 25%, eluent additive for LC-MS | AM0258 |
| Ammonium acetate, eluent additive for LC-MS | AM0259 |
| Ammonium formate, eluent additive for LC-MS | AM0320 |
| Formic acid, eluent additive for LC-MS | AC1076 |
| Trifluoroacetic acid, eluent additive for LC-MS | AC3144 |
| Triethylamine, eluent additive for LC-MS | TR0217 |

Auxiliary products

Once the daily work with the LC-MS equipment has been concluded, it is convenient to eliminate the remaining salts inside of the equipments by rinsing it with water for a period of time. After the salts elimination, it is recommended to keep a 2-propanol/water mixture in the equipment to inhibit the growth of microorganisms.

| Description | Reference |
|--|-----------|
| Formic acid solution, 10% in water, for cleaning, LC-MS | AC1075 |
| Ammonium acetate solution, 10 mmol/l in water, for cleaning, LC-MS | AM0262 |
| 2-Propanol/water, 50:50 (v/v), for cleaning, LC-MS | ME0797 |

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